



BERKELEY ANALYTICAL

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VOC Emissions from Building Products

Customer & Building Product Sample Information			
Report Certification			
Report number	991-003-01A-May0318		
Report date	May 3, 2018		
Certified by (Name/Title)	Raja S. Tannous, Laboratory Director		
Signature	Japs. Fr		
Date	May 3, 2018		
Standards			
Test method	CDPH/EHLB/Standard Method V1.2 (Sect. 01350)		
Acceptance criteria	CDPH/EHLB/Standard Method V1.2		
Modeling scenario(s)	CDPH/EHLB/Standard Method V1.2 Standard Classroom & Office		
Product type	Window Frame		
Customer Information			
Manufacturer or organization	Sika Corporation		
City/State/Country	Lyndhurst, NJ USA		
Contact name/Title	Steve Rosenberg, Senior VP Risk, Quality & Sustainability		
Phone number	201-508-6655		
Product Sample Information*			
Manufacturer (if not customer)	Same as above		
Product name / Number	Sikaflex 2C NS EZ Mix Limestone / 187658		
Product CSI category	Joint Sealants (07 92 00)		
Customer sample ID	not provided		
Manufacturing location	Sika Corporation, Marion, Ohio		
Date sample manufactured	Apr 3, 2018		
Date sample collected	Apr 3, 2018		
Date sample shipped	Apr 4, 2018		
Date sample received by lab	Apr 9, 2018		
Condition of received sample	No observed problems		
Lab sample tracking number	991-003-01A		
Conditioning start date & duration	Apr 13, 2018; 10 days		
Chamber test start date & duration	Apr 23, 2018; 4 days (96 hours)		
Total test start date & duration	Apr 13, 2018; 14 days (336 hours)		

*Chain-of-custody (COC) form for product sample is attached to this report





Conformity Assessment – CDPH VOC Concentration Criteria

VOC Emission Test Results – The product sample was tested for emissions of VOCs following California Department of Public Health CDPH/EHLB/Standard Method Version 1.2, 2017. The chamber test results were modeled to one or more scenario(s) defined in CDPH Standard Method V1.2. The modeled indoor VOC concentrations then were compared to the acceptance criteria defined in CDPH Standard Method V1.2 to determine compliance of the product sample to the standard. The modeling scenario(s) are detailed in Table 3, and the predicted indoor VOC concentrations at 336 hours are given in Table 6 of this report. The allowable concentrations used as acceptance criteria are reproduced in Appendix B of this report. Table 1 summarizes the pass/fail results based on the predicted indoor air concentrations of individual VOCs of concern in the modeled scenario(s).

TVOC Concentration Range – USGBC's LEED v4 rating systems for buildings include a requirement for reporting of the predicted TVOC concentration in one of three range categories, i.e., $\leq 0.5 \text{ mg/m}^3$, $>0.5 \text{ to } 4.9 \text{mg/m}^3$, and $\geq 5.0 \text{ mg/m}^3$. Table 1 includes the TVOC concentration range in the modeled scenario(s).

Chemical	CAS No	Allowable Concentration		oncentration /Fail)
		(µg/m³)	Classroom	Office
Ethylbenzene	100-41-4	1000	Pass	Pass
Xylenes, technical mixture (m-, p-, o- xylene combined)	1330-20-7	350	Pass	Pass
TVOC ^a			≤ 0.5 mg/m ³	≤ 0.5 mg/m

 Table 1. Pass/Fail results based on the test method and identified modeling scenarios. Only detected individual VOCs with defined acceptance criteria are listed. The TVOC concentration range also is shown

^a Reporting of TVOC range is for information only; TVOC is not a Pass/Fail criterion





Test Method for Building Product Samples

Test Specimen Preparation – Customer provide two component polyurethane elastomeric sealant in a container. Mixed 307.24g of part A and 95.18g of part B sample per TDS mixing ratio, applied 17.19g final product sample into a 20.1cm long 1/2" wide aluminum channel. Bead size is approximate 1/2" wide and 1/4" deep per customer instruction. Exposed amount is based on length of 20.1cm. Photographs of the tested specimen are shown later in this report. The test results presented herein are specific to this item.

Test Protocol Summary* – This VOC emission test was performed following California Department of Public Health <u>CDPH/EHLB/Standard Method Version 1.2, 2017</u>. This version of the standard is identical to CDPH/EHLB/Standard Method V1.1, 2010 except that the benzene allowable concentration is lower. Note: this standard derives from California architectural Specification 01350 and frequently is referred to as "Section 01350." The chamber test prescribed in the standard follows the guidance of <u>ASTM Standard Guide</u>

<u>D 5116</u>. Chemical sampling and analyses were performed following <u>U.S. EPA Compendium Method TO-17</u> and <u>ASTM Standard Method D 5197</u>. The product specimen was prepared from the supplied product sample and was placed directly into the conditioning environment and maintained at controlled conditions of air flow rate, temperature and relative humidity for ten days. At the end of this period, the specimen was transferred directly to a small-scale chamber. The chamber conditions for the 96-h test period are summarized in Table 2. Air samples were collected from the chamber at 24 h, 48 h and 96 h elapsed time. Samples for the analysis of individual VOCs and TVOC were collected on multisorbent tubes containing Tenax-TA backed by a carbonaceous sorbent. Samples for the analysis of low molecular weight aldehydes were collected on treated DNPH cartridges. VOC samples were analyzed by thermal desorption GC/MS. TVOC was calculated using toluene as the calibration reference. Individual VOCs (iVOCs) were quantified using multi-point (4 or more points) with calibration curves prepared with pure standards, unless otherwise noted. iVOCs without pure standards were quantified based on their total-ion-current responses using toluene as the calibration reference. Formaldehyde and acetaldehyde were analyzed by HPLC and quantified using multi-point (4 or more points) calibration curves. The analytical instruments and their operating parameters are described in Appendix A.

Availability of Data – All data, including but not limited to raw instrument files, calibration files, and quality control checks used to generate the test results will be made available to the customer upon request.

Parameter	Symbol	Units	Value
Tested specimen exposed area	Ls	m	0.201
Chamber volume	Vc	m ³	0.067
Loading ratio	L	m/m ³	3
Avg. Inlet gas flow rate & Range	Qc	m³/h	0.067 (0.064-0.070)
Avg Temperature & Range		°C	22.9 (22-24)
Avg Relative humidity & Range		%	50 (45-55)
Duration		h	96

 Table 2. Chamber conditions for test period

^{*}All standards identified in this section are included in Berkeley Analytical's scope of ISO/IEC17025 accreditation, Testing Laboratory TL-383, International Accreditation Service, www.iasonline.org





Modeling Parameters for Building Products

Modeling Parameters – CDPH/EHLB/Standard Method Version 1.2 describes the modeling procedures and parameters for estimating the impact of VOC emissions from a building product on indoor air concentrations in a standard classroom and a standard office space. The dimensions and ventilation of the spaces and the exposed surface areas of major materials are prescribed. The modeling scenario(s) and parameters applicable to this test are listed in Table 3.

Table 3.	Parameters used for	estimating VOC a	ir concentrations at 33	6 hours for the m	odeling scenarios
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Parameter	Symbol	Units	Value		
Falameter	Symbol	Onits	Classroom	Office	
Product exposed area	L _{PB}	m	12.2	4.883	
Building volume	VB	m³	231	30.6	
Floor/Ceiling Area	AB	m²	89.2	11.15	
Ceiling height	Нв	m	2.59	2.74	
Outdoor air (OA) flow rate	QB	m³/h	191	20.7	
Area-specific air flow rate	QA	m³/m-h	15.66	4.23	





VOC Emission Test Results

Chamber Background Concentrations – Background concentrations measured at time zero are reported in Table 4. The background concentrations of TVOC, formaldehyde, acetaldehyde, and reported iVOCs are listed.

Chemical/Chemical Group	CAS No	Chamber Conc (µg/m³)
2-Propanone (acetone)	67-64-1	3.8
Acetaldehyde	75-07-0	3.5
Formaldehyde	50-00-0	LQ
TVOC		LQ

Table 4. Chamber background VOC concentrations at time zero

Emitted VOCs – Individual VOCs (iVOCs) detected in the test and present above the lower limits of quantitation in chamber air are reported in Table 5. All iVOCs with CRELs and/or on other lists of toxicants of concern are listed first. Next, all frequently occurring iVOCs with pure standard calibrations are listed. Additionally, the 10 most abundant iVOCs quantified using toluene as the reference standard are listed; identifications of these compounds are considered tentative. Reporting of fewer than 10 iVOCs indicates that fewer than 10 chemicals met these criteria.

Table 5. Listed and abundant iVOCs detected above lower limits of quantitation in 96-h air sample

Chemical	CAS No	Surrogate?*	CREL (µg/m³)	CARB TAC Category	Prop 65 List?
Ethylbenzene	100-41-4		2000	T-lla	Yes
Xylenes, technical mixture (m-, p-, o- xylene combined)	1330-20-7		700	T-Ila	
Isopropylbenzene (cumene)	98-82-8			T-IVa	Yes
2-Propanone (acetone)	67-64-1				
Propylene glycol methyl ether acetate	108-65-6				

*"Yes" response indicates iVOC quantified using toluene as the calibration reference; all other iVOCs quantified using pure standards



VOC Emission Test Results, Continued

VOC Emission Factors and Estimated Indoor Air Concentrations – The 96-h chamber sample was analyzed for iVOCs including formaldehyde and acetaldehyde. The emission factors for iVOCs presented in Table 6 were calculated from the chamber parameters, the exposed area of the test specimen and the measured 96-h chamber concentrations corrected for any chamber background concentrations. The emission factors were used to predict the indoor air concentrations of iVOCs for the modeling scenario(s) applicable to this test as shown in Table 3. See Equations for calculation methods.

Chemical	Chamber Concentration	Emission Factor	Estimated Indoor Air Concentration (µg/m ³)	
	(µg/m³)	(µg/m²-h)	Classroom	Office
2-Propanone (acetone)	10.6	3.5	0.2	0.8
Ethylbenzene	993.5	332.6	21.2	78.6
Xylenes, technical mixture (m-, p-, o- xylene combined)	3290.4	1101.5	70.3	260.4
Propylene glycol methyl ether acetate	448.0	150.0	9.6	35.4
Isopropylbenzene (cumene)	6.6	2.2	0.1	0.5

Table 6. Measured chamber concentrations at 96 h, calculated emission factors, and estimated indoor air concentrations of individual VOCs for the modeling scenarios





VOC Emission Test Results, Continued

Quality Measurements – Chamber samples collected at 24, 48 and 96 hours were analyzed for total VOCs (TVOC). Because the TVOC response per unit mass of a chemical is highly dependent upon the specific mixture of iVOCs, the measurement of TVOC is semi-quantitative. TVOC primarily is used as a quality measure to determine if the VOC emissions from a product are relatively constant or generally declining over the test period. Some programs may require the reporting of predicted indoor air TVOC concentrations or concentration ranges in mg/m³. TVOC emission factors and predicted TVOC concentrations are shown in Table 7. Aldehyde samples collected at 24, 48 and 96 hours were analyzed for formaldehyde as another quality measure. Formaldehyde emission factors are shown in Table 8. Product claims related to formaldehyde content may be based, in part, on formaldehyde emission factors.

 Table 7. TVOC chamber concentrations at 24, 48, and 96 h with corresponding emission factors and predicted indoor air concentrations (mg/m³)

Elapsed Time	Chamber Concentration	Emission Factor	Estimated Indoor A (mg/r	
(h)	(µg/m³)	(µg/m²-h)	Classroom	Office
24	2697	903	0.058	0.213
48	3387	1134	0.072	0.268
96	4940	1654	0.106	0.391

Table 8. Formaldehyde chamber concentrations at 24, 48, and 96 h with corresponding emission factors

Elapsed Time (h)	Chamber Concentration (μg/m³)	Emission Factor (µg/m²-h)
24	LQ	LQ
48	LQ	LQ
96	LQ	LQ





Photographs of Tested Product Specimen

Photo Documentation – The product sample specimen is photographed immediately following specimen preparation and prior to initiating the conditioning period. Typically, the top and bottom faces of the specimen are photographed. Bottom faces may show a stainless steel plate or other substrate if prescribed by the standard.





Definitions, Equations, and Comments

Table 9. Definitions of parameters

Parameter/Value	Definition		
CARB TAC	Toxic Air Contaminant (TAC) on California Air Resources Board list, with toxic category indicated		
CAS No.	Chemical Abstract Service registry number providing unique chemical ID		
Chamber Conc.	Measured chamber VOC concentration at time point minus any analytical blank or background concentration for empty chamber measured prior to test. Lower limit of quantitation (LQ) or reporting limit for individual VOCs is 2 μ g/m ³ unless otherwise noted		
Indoor Air Conc.	Estimated indoor air concentration in standard modeled environment calculated from the emission factors from test results and the modeling parameters in Table 3 using the equations given below		
CREL	Chronic non-cancer Reference Exposure Level established by Cal/EPA OEHHA (http://www.OEHHA.ca.gov/air/allrels.html)		
Emission Factor	Mass of compound emitted per unit area per hour (calculation shown below). Reporting limits for emission factors are established by LQ or reporting limit for chamber concentration and specimen area tested		
Formaldehyde & acetaldehyde	Volatile aldehydes quantified by HPLC following ASTM Standard Method D 5197. LQs for formaldehyde and acetaldehyde are 1.1 µg/m ³ and 2 µg/m ³ , respectively		
Individual VOCs	Quantified by thermal desorption GC/MS following EPA Method TO-17. Compounds quantified using multi-point calibrations prepared with pure chemicals unless otherwise indicated. VOCs with chronic RELs are listed first, followed by other TAC and Prop. 65 compounds. Additional abundant VOCs at or above reporting limit of 2 µg/m ³ are listed last		
LQ	Indicates calculated value is below its lower limit of quantitation		
Prop 65 list	"Yes" indicates the compound is a chemical known to cause cancer or reproductive toxicity according to California Safe Drinking Water Toxic Enforcement Act of 1986 (Proposition 65)		
TVOC	Total Volatile Organic Compounds eluting over retention time range bounded by n-pentane and n-heptadecane and quantified by GC/MS TIC method using toluene as calibration reference. LQ for TVOC is 20 µg/m ³		
"na"	Not applicable		
"<"	Less than value established by LQ		

Equations Used in Calculations – An emission factor (EF) in μ g/m²-h for a chemical in a chamber test of a building product sample is calculated using Equation 1:

$$EF = (Q_c (C - C_o)) / A_s$$
 (1)

where Q_c is the chamber inlet air flow rate (m³/h), C is the VOC chamber concentration ($\mu g/m^3$), C₀ is the corresponding chamber background VOC concentration ($\mu g/m^3$), and A_s is the tested specimen exposed area (m²).





Definitions, Equations, and Comments, Continued

The indoor air concentration (C_B) for the modeled space in $\mu g/m^3$ is estimated using Equation 2 and the parameters defined in Table 3:

$$C_{\rm B} = (EF \times A_{\rm P_{\rm B}}) / Q_{\rm B}$$
 (2)

where A_{P_B} is the exposed area of the product in the building (m²) and Q_B is the outside air flow rate (m³/h).

Comments: Treated as window perimeter sealant.

END OF REPORT





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Appendix A Analytical Instruments & Operating Parameters

Table A1. Description of analytical instrument components

Component	Description
HPLC	1260 Infinity Quaternary LC, G1314F VW Detector, Agilent
Analytical column	Poroshell 120 EC-C18, Agilent
Column dimensions	2.1 mm x 100 mm
Thermal desorber	Unity / TD100, Markes International, Ltd.
Gas chromatograph	Model 7890A, Agilent
Analytical column	DB-624, J&W Scientific
Column dimensions	1 μm film, 0.18 mm ID, 20 m
Mass spectrometer	Model 5975C MSD, Agilent

 Table A2. HPLC operating parameters for analysis of formaldehyde and acetaldehyde

Parameter	Value
Solvent A	65/35% H₂O/Acetonitrile
Solvent B	100% Acetonitrile
Flow rate	0.3 mL/min
End time	11 min
Detector wavelength	360 nm

 Table A3.
 Thermal desorption GC/MS parameters used for analysis of iVOCs and TVOC

Parameter	Value
Thermal desorption	
Tube desorb temperature	285 °C
Trap temperature	-5 °C
Trap desorb temperature	300 °C
Trap desorb split ratio	10:1
Gas chromatograph	
Initial temperature	40°C
Initial temperature time	6.0 min
Final temperature	225 °C
Final temperature time	3 min
Mass spectrometer	
Low scan mass, <i>m/z</i>	30 amu
High scan mass, <i>m/z</i>	450 amu
Scan rate	3.42 Hz





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Appendix B Target CREL VOCs and Their Maximum Allowable Concentrations Copied from CDPH/EHLB/Standard Method Version 1.2, 2017, Table 4-1

No.	Compound Name	CAS No.	Allowable Conc. (µg/m ³)
1	Acetaldehyde	75-07-0	70
2	Benzene	71-43-2	1.5
3	Carbon disulfide	75-15-0	400
4	Carbon tetrachloride	56-23-5	20
5	Chlorobenzene	108-90-7	500
6	Chloroform	67-66-3	150
7	Dichlorobenzene (1,4-)	106-46-7	400
8	Dichloroethylene (1,1)	75-35-4	35
9	Dimethylformamide (N,N-)	68-12-2	40
10	Dioxane (1,4-)	123-91-1	1,500
11	Epichlorohydrin	106-89-8	1.5
12	Ethylbenzene	100-41-4	1,000
13	Ethylene glycol	107-21-1	200
14	Ethylene glycol monoethyl ether	110-80-5	35
15	Ethylene glycol monoethyl ether acetate	111-15-9	150
16	Ethylene glycol monomethyl ether	109-86-4	30
17	Ethylene glycol monomethyl ether acetate	110-49-6	45
18	Formaldehyde	50-00-0	9*
19	Hexane (n-)	110-54-3	3,500
20	Isophorone	78-59-1	1,000
21	Isopropanol	67-63-0	3,500
22	Methyl chloroform	71-55-6	500
23	Methylene chloride	75-09-2	200
24	Methyl t-butyl ether	1634-04-4	4,000
25	Naphthalene	91-20-3	4.5
26	Phenol	108-95-2	100
27	Propylene glycol monomethyl ether	107-98-2	3,500
28	Styrene	100-42-5	450
29	Tetrachloroethylene	127-18-4	17.5
30	Toluene	108-88-3	150
31	Trichloroethylene	79-01-6	300
32	Vinyl acetate	108-05-4	100
33-35	Xylenes, technical mixture	108-38-3,	350
	(m-, o-, and p- xylene combined)	95-47-6,	
		106-42-3	

*All maximum allowable concentrations are one half the corresponding CREL adopted by Cal/EPA OEHHA with the exception of formaldehyde for which the full CREL of 9 μ g/m³ is allowed.

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Customer Information*

Company: Sika Corporation Street Address: 201 Polito Avenue City/State/Zip(postal code): Lyndhurst, NJ 07876 Country: USA Contact Name & Title (for reporting): Steve Rosenberg, Senior VP Risk, Quality & Sustair Contact Phone/Fax Numbers: 201 508 6655 / 201 933 6134 Contact E-mail Address: rosenberg.steve@us.sika.com Financially Responsible Co. (if different):

Manufacturer Information (if different from customer)

- Company: Sika Corporation
- City/State/Country: Marion, OH USA
- Contact Name/Title: Matt Fuller/Quality Lab Supervisor
- Phone Number/E-mail Address: 740-375-3069/fuller.matt@us.sika.com

Sample Details Product Commercial Name*: Sikaflex 2C NS EZ Mix Limestone Product Commercial Part No.(if not part of the name)* 187658 Manufacturer Sample Tracking ID: Date Manufactured*: 4/3/2018 Product Category & Use*: None sagging Construction Sealant Sample Construction Material*: Pail A/B-Lot Number 3003210299 Plant Name & Location*: Sika Corporation, Marion, Ohio Collection Location within Plant: M41 Date & Time Collected* : 4/3/2018 time-1430 Photo(s) of Collection Location: Attach Number of Sample Pieces*: 1 Sample Collected by*: Matt Fuller Phone/Fax Numbers*: same as above E-mail Address*: same as above Shipping Details*

Packed & Shipped By: KOB LOE Shipping Date: 4/4/2018 Carrier/Airbill Number: 1242848X5351794775

Chain of Custody for Building Product/ Material VOC Emission Test

A Separate COC must be completed for EACH product/material sample

A link to Berkeley Analytical's Terms & Conditions is included in this workbook. By submitting samples,

customer acknowledges and accepts these terms & conditions unless a prior written contract is in effect.

Berkeley Analytical Quotation Number: 170821-1

Purchase Order (enter company & number): 4501407517

Requested Test (automatically fill	ed from BldgProdWorksheet Selections)	
Test to be performed	CDPH Std. Method V1.1	
Modeling scenario	Office & Classroom	
Test schedule (screening tests only)		
Target chemicals and chemical groups (screening)		
CARB ATCM test, schedule		
Test results application(s)	LEED,	
For Berkeley Analytical Use:		
Report ID	RPT66	
Billing Reference		
Quetemes Instructions for Sample Bran Tool	t Type schedule etc. (filled from BldProdWorksheet	

Customer Instructions for Sample Prep., Test Type, schedule, etc. (filled from BidProdWorksheet)

Sikaflex 2C NS EZ Mix Limestone is a 2 component none sagging construction sealant and should be applied at 1/2 inch wide by 1/4 inch deep bead on the perimeter of windows.

Customer Request for Laboratory Certificate of Compliance

Indicate if you are ordering a Laboratory Certificate of Compliance: Yes

Laboratory certificates are available for the compliance test(s) listed on the BldgProdWorksheet. Berkeley Analytical's laboratory test results and associated certificates are specific to the tested item. Claims made by the customer regarding the broader representativeness of the test results and certificate are the sole responsibility of the customer.

Customer Authorizes Laboratory to Submit Copies of Test Report to:

Contact/E-mail Address:

Organization:

Contact/E-mail Address:

Organization:

F	or Berkeley Analytical Use Only	
Condition of Shipping Package:	ok	
Condition of Sample:	ok	
Lab Tracking Number: 99	1-003-01 A	i)

Sample Handling				
Relinquished By*	Received By*	Signature*	Date*	Company*
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© Copyright, Berkeley Analytical Associates, LLC, Aug 2013	ALECTUANG	Alecturo	4-9-18 BKA	



BUILDING TRUST

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Sent by e mail

SCENARIO FOR USE OF SIKAFLEX 2C NS EZ MIX ACCORDING TO CPDH IAQ STANDARD

Lyndhurst, 3 April 2018

Dear Al:

I have reviewed the various scenarios in the CPDH IAQ standard for a standard school classroom and a standard private office. The following are the guidelines for the use of Sikaflex 2c NS EZ.

For both the School and the Office Sikaflex 2C NS EZ would be applied as a perimeter seal for the window. The bead size would most likely be 1/2 wide by 1/4 inch deep.

Attached please find the Product Data Sheet for Sikaflex 2C NS EZ along with SDS for the Part A and Part B. As this is a two component material the proper mix ratio is 3 parts by weight of the Part A to 1 Part by weight of the Part B. The actual material you will be receiving is Sikaflex 2C NS EZ Limestone as this has the pigment added into the Part A.

If you have any questions or need further information please feel free to contact me.

Sincerely,

Steven A. Rosenberg Enclosure

SIKA CORPORATION USA